## 1,2-HOMOHEPTAFULVENE: SYNTHESIS AND CYCLOADDITION

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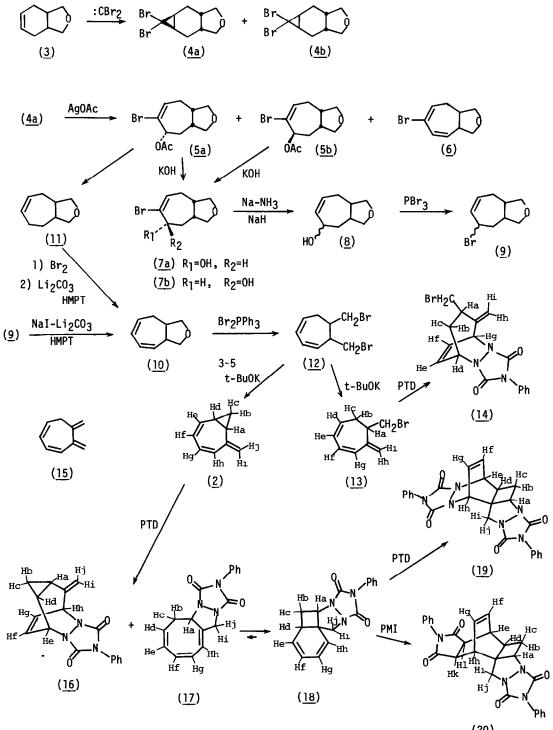
Summary: 1,2-Homoheptafulvene (2) was synthesized and reacted with 4-phenyl-1,2,4-triazoline-3,5-dione to yield  $[4\pi + 2\pi]$  and  $[6\pi + 2\sigma + 2\pi]$  cycloadducts.

Heptafulvene (<u>1</u>) has been synthesized by Doering<sup>1</sup>) and later by Zimmerman<sup>2</sup>) and by Neuenschwander<sup>3</sup>) as an unstable polyolefinic hydrocarbon, and has been know to react with dienophiles to yield  $[8\pi + 2\pi]$  cycloadducts.<sup>1, 4</sup>) However, 1,2-homoheptafulvene (<u>2</u>) is unknown compound although 8,8-disubstituted derivatives such as 8,8-dicyano- and 8,8-dipheny1-1,2-homoheptafulvenes have recently been synthesized.<sup>5</sup>)



We succeeded to synthesize 1,2-homoheptafulvene starting from 8-oxabicyclo[4.3.0]-3-nonene (3), and found that it reacted with dienophile to give cycloadducts by the modes of  $[4\pi + 2\pi]$  and  $[6\pi + 2\sigma + 2\pi]$ . The results are reported in this paper.

The reaction of <u>3</u> with large excess of dibromocarbene generated from bromoform and t-BuOK in hexane afforded two stereoisomeric adducts (<u>4a</u>, 70-80%, mp 98.5~100.5°) and (<u>4b</u>, 2%, mp 41.5~ 42.5°).<sup>6</sup>, <sup>7</sup>) Treatment of the major adduct (<u>4a</u>) with silver acetate in refluxing acetic acid gave ring enlargement products (<u>5a</u>, 61%, mp 98.5~100.5°), (<u>5b</u>, 36%, mp 65~66°) and (<u>6</u>, 2%, oil).<sup>8</sup> Hydrolyses of <u>5a</u> and <u>5b</u> with alcoholic KOH afforded the corresponding bromoalcohols (<u>7a</u>, mp 89~90°) and (<u>7b</u>, mp 64~65°) in 95% yields, respectively. Birch reduction of a mixture of the bromoalcohol with sodium in liquid ammonia in the presence of sodium hydride afforded a



(<u>20</u>)

stereoisomeric mixture of alcohol ( $\underline{8}$ , 30~45%, oil bp 85~95°/5 mm), treatment of  $\underline{8}$  with PBr<sub>3</sub> in anhydrous ether yielded a bromide ( $\underline{9}$ , 50~70%, mp 58~59°), which was dehydrobrominated with NaI-Li<sub>2</sub>CO<sub>3</sub> in HMPT to afford 8-oxabicyclo[5.3.0]-2,4-diene ( $\underline{10}$ , 65~72%, oil bp 92~95°/45 mm).

Compound (<u>10</u>) was also obtained by an alternative route. Thus, Birch reduction of a mixture of <u>5a</u> and <u>5b</u> afforded monoene (<u>11</u>, 42%, oil bp 95-105°/45 mm), whose bromination in CCl<sub>4</sub> followed by dehydrobromination with  $Li_2CO_3$  in HMPT afforded the diene (<u>10</u>, 45~50%).

Reaction of <u>10</u> with  $Br_2^{PPh_3}$  in chlorobenzene at 90°C gave 5,6-bisbromomethylcyclohepta-1,3diene (<u>12</u>, 40-50%, oil). Reaction of <u>12</u> with equimolar amount of t-BuOK in THF afforded 1-bromomethyl-1,2-dihydroheptafulvene (<u>13</u>, 48%, colorless oil) accompanying a recovery of <u>12</u> (19%). The compound (<u>13</u>) reacted with 4-phenyl-1,2,4-triazoline-3,5-dione (PTD) to give  $[4\pi + 2\pi]$ cycloadduct (14, mp 178-181°).

The dehydrobromination of <u>12</u> using 3-5 molar equivalents of t-BuOK in DMSO or THF followed by chromatographic separation on silica gel using pentane as solvent yielded 1,2-homoheptafulvene (<u>2</u>, 54%) as a moderately thermally stable but air sensitive colorless oil, and 1,2-dimethylenecyclohepta-3,5-diene (<u>15</u>) did not form at all. The structure of <u>2</u> was determined by NMR analysis (Table 1) and by cycloaddition with PTD, and furthermore the compound was found to differ from <u>15</u> synthesized by other route.<sup>9</sup>) Electronic spectrum of <u>2</u> shows maxima at 215 nm ( $\varepsilon$  3,900) and 299 nm ( $\varepsilon$  2,350), and is completely different from that of heptafulvene (<u>1</u>) which shows maxima at 370-580 nm ( $\varepsilon$  100-500) with fine structure as well as at around 280 nm ( $\varepsilon$  ca. 13,000).<sup>1-3</sup>)

1,2-Homoheptafulvene (2) reacted with PTD to give 1 : 1 adducts (<u>16</u>, 32%, mp 205-210°(d)) and (<u>17</u>, 20%, mp 123-125°). The compounds (<u>16</u>) and (<u>17</u>) were determined to be  $[4\pi + 2\pi]$  and  $[6\pi + 2\sigma + 2\pi]$  cycloadducts, respectively, as shown in scheme by NMR analysis (Table 1). <sup>1</sup>H-NMR of <u>16</u> shows cyclopropane ring protons at 0.76-1.13 (2H) and 1.58-1.83 ppm (2H). However, the NMR of <u>17</u> does not show signals at higher than 2 ppm, and there appears instead a pair of signals at 2.57 and 3.00 ppm with a large coupling of J = 18.5 Hz for the hydrogen atoms which originated from the cyclopropane ring.

A solution of <u>17</u> in chloroform or benzene gradually changed to tetracyclic valence isomer (<u>18</u>) although the compound could not be obtained in pure crystalline state. The mixture of <u>17</u> and <u>18</u>, enriched with <u>18</u> by standing for a week, reacted with PTD and N-phenylmaleinimide (PMI) to give further  $[4\pi + 2\pi]$  cycloadducts (19, 66%, mp 201~209°(d)) and (<u>20</u>, 68%, mp 308~309°), respectively.

Spectroscopic data of 2, 13, 14, and  $16 \sim 20$  are shown in Table 1. Further studies on physical and chemical properties of 2 are now in progress. Table 1. Spectroscopic Data of 2, 13, 14, and 16~20. Compounds Spectroscopic Data [NMR (90 MHz); & ppm (J in Hz)] IR (CC1<sub>A</sub>); 1640, 1595, 1570, 870 cm<sup>-1</sup>, UV  $\lambda$ max (hexane); 215 nm ( $\epsilon$  3,900), 299 nm ( $\epsilon$ 2,350), NMR (CC1<sub>4</sub>); 1.2~1.6 (m, Hb,c,d), 2.05 (t,d, J=8, 8, Ha), 4.85 (d, J=2, Hi or 2 Hj), 5.08 (d, J=2, Hj or Hi), 5.3~5.6 (m, olefinic 2H), 5.8~6.2 (m, olefinic 2H) NMR (CC1<sub>A</sub>); 2.7~3.1 (m, Ha,b,c), 3.20 (d,d, J=10, 5, CHBr), 3.44 (t, J=10, CHBr), 4.99 13 (bs, Hh or Hi), 5.03 (bs, Hi or Hh), 5.55~6.0 (m, Hd,e,f), 6.11 (bd, J=11, Hg) NMR (CDCl<sub>3</sub>); 1.88 (d,d,d, J=12.5, 11, 2, Hb), 2.41 (d,d,d, J=12.5, 6, 6, Hc), 2.72 (m, <u>14</u> Ha), 3.61 (m, CH<sub>2</sub>Br), 5.11 (d, J=1, Hh or Hi), 5.19 (d, J=1, Hi or Hh), 5.33 (d,d, J= 4.5, 3, Hd), 5.40 (bs, Hg), 6.39 (d,d,m, J=5, 4.5, He,f), 7.2~7.6 (m, aromatic 5H) NMR (CDC1<sub>2</sub>); 0.76~1.13 (m, Hb,c), 1.58~1.83 (m, Ha,d), 5.07 (bd, J=6.5, He), 5.11 (bs, Hi or Hj), 5.18 (bs, Hj or Hi), 5.39 (m, Hh), 6.04 (d,d,d, J=8.5, 6.5, 1, Hf or Hg), <u>16</u> 6.26 (d,d,d, J=8.5, 6.5, 1, Hg or Hf), 7.3-7.6 (m, aromatic 5H) NMR (CDC1<sub>2</sub>); 2.57 (d,d,d, J=18.5, 11, 4, Hb), 3.00 (d,d,d, J=18.5, 4, 2, Hc), 4.2617 (d,d,d, J=14, 2.5, 1, Hi), 4.61 (d,d, J=14, 1.5, Hj), 5.13 (d,d, J=11, 4, Ha), 5.75~6.2 (m, olefinic 5H), 7.2~7.6 (m, aromatic 5H) NMR (CDC1<sub>3</sub>); 2.65 (m, Hb,c), 3.1 (m, Ha), 3.24 (d, J=10, Hi), 3.95 (d, J=10, Hj), 4.85 <u>18</u> (m, Hd), 5.50 (bd, J=9, Hh), 5.55~6.2 (m, He,f,g), 7.2~7.6 (m, aromatic 5H) NMR (Py- $d_5$ ); 1.76 (d,t, J=14, 6.5, Hb or Hc), 2.20 (d,d,d, J=14, 9, 4, Hc or Hb), 2.94 (m, Hd), 3.91 (d, J=11, Hi), 4.18 (d, J=11, Hj), 4.25 (d,d, J=6.5, 4, Ha), 5.04 (m, <u>19</u> He,h), 6.53 (m, Hf,g), 7.2~7.6 (m, aromatic 10H) NMR (Py- $d_5$ ); 1.76 (d,t, J=13.5, 6.7, Hb or Hc), 2.14 (d,d,d, J=13.5, 8.6, 3.6, Hc or Hb), 2.56 (m, Hd), 3.22 (bs, Hk,1), 3.42 (m, He,h), 3.92 (d, J=10, Hi), 4.17 (d, J=10, 20 Hj), 4.25 (d,d, J=6.7, 3.6, Ha), 6.38 (m, Hf,g), 7.25~7.6 (m, aromatic 10H) References 1) W. von E. Doering and D. W. Wiley, Tetrahedron, 11, 183 (1960). 2) H. E. Zimmerman and L. R. Sousa, J. Am. Chem. Soc., 94, 834 (1972). 3) M. Neuenschwander and W. K. Schenk, Chimia, 26, 194 (1972); W. K. Schenk, R. Kyburz, and

- M. Neuenschwander, Helv. Chim. Acta, <u>58</u>, 1099 (1975).
- 4) R. B. Woodward and R. Hofmann, Angew. Chem., <u>81</u>, 797 (1969).
- 5) M. Oda, private communication.
- 6) All new compounds gave satisfactory elemental analyses and/or correct mass spectra.
- The stereochemistry was determined by NMR analysis using shift reagent.
- 8) The stereochemistry was tentatively assigned by NMR analysis using shift reagent.

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9) M. Oda, N. Morita, and T. Asao, to be published.